

**Supplementary material for: Density functional theory calculations
of mechanical and electronic properties of $W_{1-x}Ta_xN_6$, $W_{1-x}Mo_xN_6$,
and $Mo_{1-x}Ta_xN_6$ ($0 \leq x \leq 1$) alloys in hexagonal structure**

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Table S1: Lattice constants (a , c), elastic constants (C_{11} , C_{12} , C_{13} , C_{33} , and C_{44}), charge transfer per atom from metal to nitrogen (Q_t) in units of elementary charge e , and total magnetic moment (Mag.) in Bohr magneton (μ_B) per transition metal atom for pure MN_6 compounds¹.

Compounds	a (Å)	c (Å)	C_{11} (GPa)	C_{12} (GPa)	C_{13} (GPa)	C_{33} (GPa)	C_{44} (GPa)	Q_t (e)	Mag. (μ_B)
WN ₆	6.16	4.57	724	82	148	814	389	1.03	0
MoN ₆	6.18	4.58	677	99	145	736	354	1.09	0
TaN ₆	6.22	4.70	656	113	165	681	312	0.77	0

Table S2: Bulk modulus (B), shear modulus (G), Young's modulus (E), Vickers hardness (H_V), Pugh's ratio (k), Poisson's ratio (ν), Debye temperature (θ_D), formation energy per atom (E_f), and mass density (ρ) for pure MN_6 compounds¹.

Compounds	B (GPa)	G (GPa)	E (GPa)	H_V (GPa)	k	ν	θ_D (K)	E_f (eV)	ρ (kg/m ³)
WN ₆	333	341	762	59	1.02	0.12	1048	1.03	8889
MoN ₆	317	309	699	52	0.97	0.13	1219	1.09	5928
TaN ₆	319	280	649	43	0.88	0.16	966	0.77	8381

Fig. S1: Relation between θ_D and a hardness parameter for all $M_{1-x}M_2_xN_6$ ($0 \leq x \leq 1$) compounds.

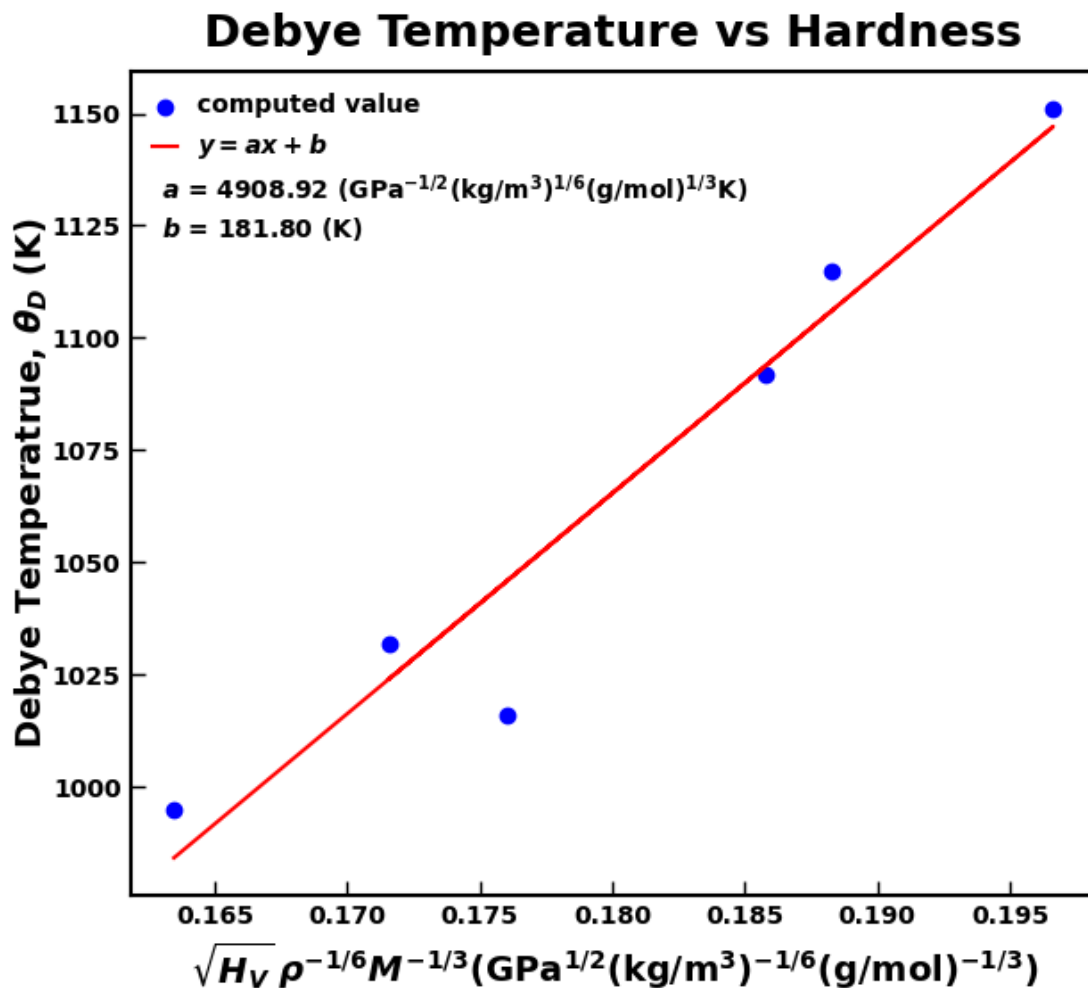
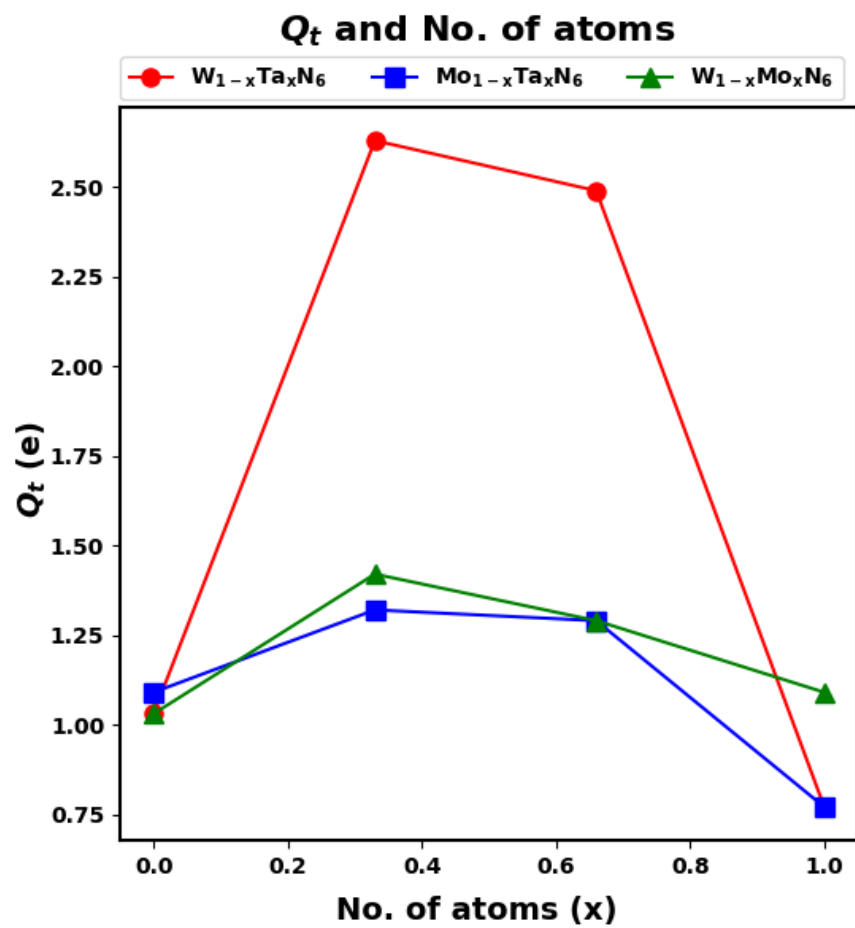


Fig. S2: Relation between Q_t and number of atoms for all $M1_{1-x}M2_xN_6$ ($0 \leq x \leq 1$) compounds.



References

1. S. R. Kandel, B. B. Dumre, D. Gall and S. V. Khare, *Materialia* **25**, 101550 (2022).