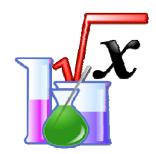
# A. Experimental Practices, andB. Graphing and Analysis with Igor Pro

Week of Sept. 6, 2010

#### IGOR Pro

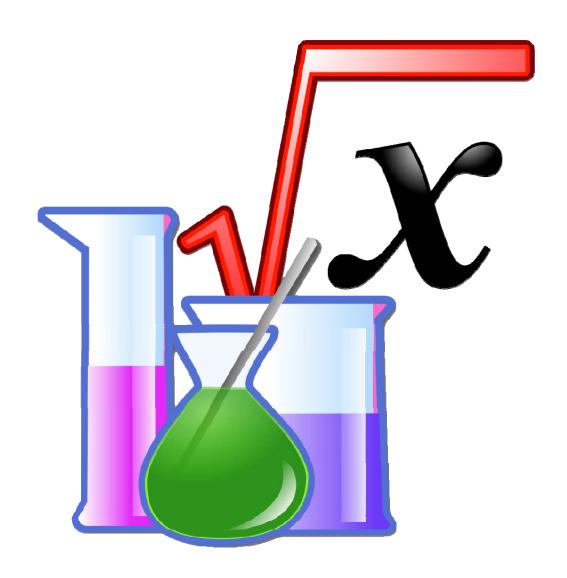
- Runs on Mac OS X and Windows
- Fast Display of Large Data Sets
- Interactive Data Exploration
- Journal—Quality Graphics
- Powerful Curve Fitting
- Extensive Data Analysis & Statistics
- Image Processing
- Data Acquisition Support
- Built-In Programming Environment Supports Analysis and Automation
- Customizable User Interface
- Used by Scientists and Engineers Worldwide Since 1989



# Atomic and Nuclear Physics Laboratory (Physics 4780)

The University of Toledo Instructor: Randy Ellingson

# **Experimental Methods and Practices**



#### **Experimental Methods and Practices**

#### From sciencebuddies.org:

#### **Key Info**

The scientific method is a way to ask and answer scientific questions by making observations and doing experiments.

The steps of the scientific method are to:

- Ask a Question
- Do Background Research
- Construct a Hypothesis
- Test Your Hypothesis by Doing an Experiment
- Analyze Your Data and Draw a Conclusion
- Communicate Your Results

It is important for your experiment to be a fair test. A "fair test" occurs when you change only one factor (variable) and keep all other conditions the same.

http://www.sciencebuddies.org/science-fair-projects/project\_scientific\_method.shtml

# **Key Elements of the Scientific Method**

- 1. Define the question
- 2. Gather information and resources (observe)
- 3. Form hypothesis
- 4. Perform experiment and collect data
- 5. Analyze data
- 6. Interpret data and draw conclusions that serve as a starting point for new hypothesis
- 7. Publish results
- 8. Retest (frequently done by other scientists)

The iterative cycle inherent in this step-by-step methodology goes from point 3 to 6 back to 3 again.

# Thoughts on the Scientific Method

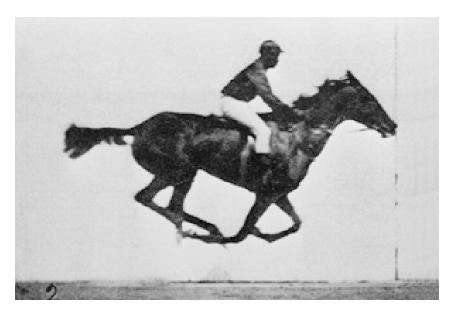
Ibn al-Haytham (Alhazen, 965–1039), pointed out the emphasis on seeking truth:

Truth is sought for its own sake. And those who are engaged upon the quest for anything for its own sake are not interested in other things. Finding the truth is difficult, and the road to it is rough.

According to William Whewell (1794–1866), "invention, sagacity, genius" are required at every step in scientific method. It is not enough to base scientific method on experience alone; multiple steps are needed in scientific method, ranging from our experience to our imagination, back and forth.

# Scientific Method: Beliefs and Biases

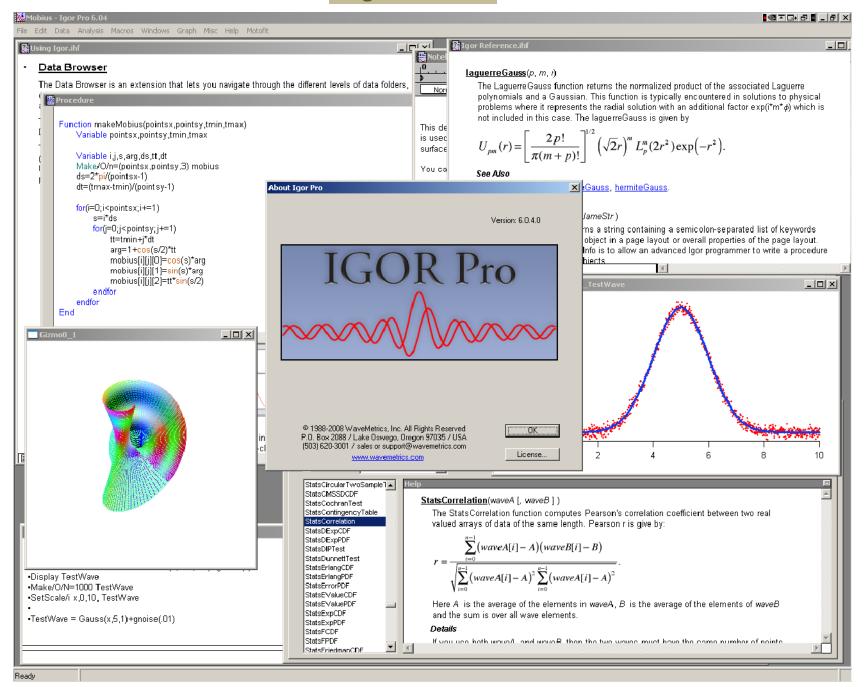
Eadweard Muybridge's (1830 – 1904) studies of a horse galloping



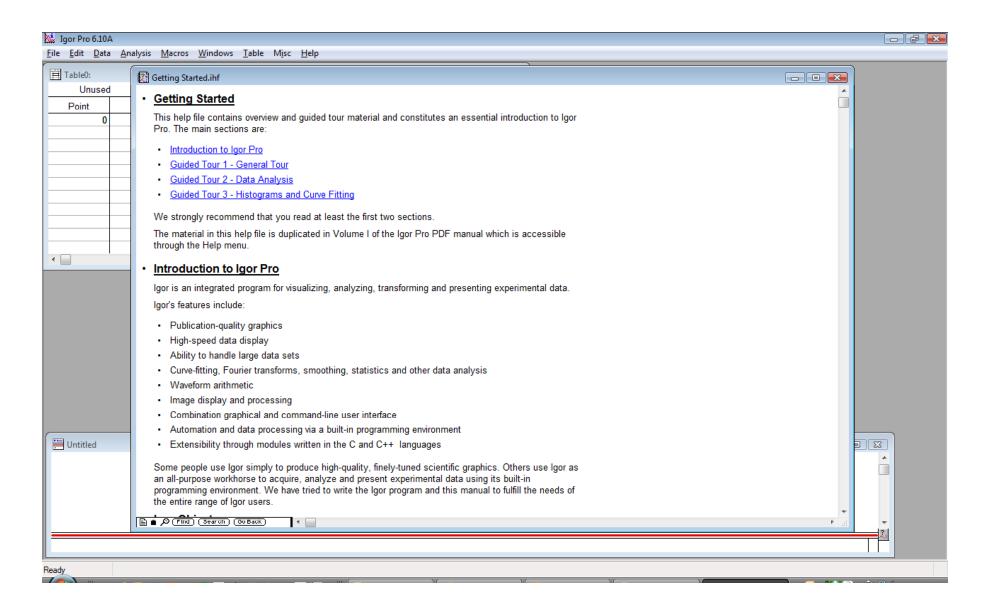


The Epsom Derby (1821) by Géricault, Jean Louis Théodore. Oil on canvas.

#### Igor Pro 6.1



# Igor Pro 6.1 (www.wavemetrics.com)



#### Error handling and propagation

- See handout (to be emailed)
- See also:

http://teacher.pas.rochester.edu/PHY\_LABS/AppendixB/AppendixB.html

### **Error (uncertainty) analysis**

$$z = f(x, y)$$

$$\Delta z = \frac{\partial f(x, y)}{\partial x} \Delta x + \frac{\partial f(x, y)}{\partial y} \Delta y$$

For uncorrelated errors:

$$\Delta z = \left[ \left( \frac{\partial f(x, y)}{\partial x} \Delta x \right)^2 + \left( \frac{\partial f(x, y)}{\partial y} \Delta y \right)^2 \right]^{\frac{1}{2}}$$

#### Error (uncertainty) analysis – example using Activity of <sup>241</sup>Am source

$$A = \frac{C}{\pi s^2 / 4\pi r^2} = \frac{C(4r^2)}{s^2}$$

Where *C* is count rate, *r* is the distance to the detector, and *s* is the radius of the detector.

$$\Delta A = \frac{\partial}{\partial C} \left( \frac{4Cr^2}{s^2} \right) \Delta C + \frac{\partial}{\partial r} \left( \frac{4Cr^2}{s^2} \right) \Delta r$$

$$\Delta A = \frac{4r^2}{s^2} \Delta C + \left( \frac{4C}{s^2} \right) (2r) \Delta r$$

In quadrature, if these errors are uncorrelated:

$$\Delta A = \left[ \left( \frac{4r^2}{s^2} \Delta C \right)^2 + \left( \frac{8Cr}{s^2} \Delta r \right)^2 \right]^{\frac{1}{2}}$$

#### Error (uncertainty) analysis – multiple value r.m.s. approach

Another approach to evaluating the uncertainty relies on a straightforward calculation of the root mean square and the standard deviation. In the case of your Activity measurements, you could compute the RMS value as well as the SD. You should still assess your uncertainty through error propagation, as the uncertainty may very well exceed the SD.

Definition of the root mean square from Wikipedia:

"...the <u>square root</u> of the <u>arithmetic mean</u> (<u>average</u>) of the <u>squares</u> of the original values..."

In the case of a set of *n* values  $x_1$ ,  $x_2$ , ...  $x_n$ , the RMS value is given by:

$$x_{rms} = \sqrt{\frac{x_1^2 + x_2^2 + \dots + x_n^2}{n}}$$

The **standard deviation** is given as follows, where there are N values and  $\mu$  is the arithmetic mean:

$$\sigma = \sqrt{\frac{1}{N}} \sum_{i=1}^{N} (x_i - \mu)^2$$