



Full paper

Compositional and morphological engineering of mixed cation perovskite films for highly efficient planar and flexible solar cells with reduced hysteresis



Changlei Wang^{a,b}, Dewei Zhao^{b,*}, Yue Yu^b, Niraj Shrestha^b, Corey R. Grice^b, Weiqiang Liao^b, Alexander J. Cimaroli^b, Jing Chen^c, Randy J. Ellingson^b, Xingzhong Zhao^{a,*}, Yanfa Yan^{b,*}

^a Key Laboratory of Artificial Micro/Nano Structures of Ministry of Education, School of Physics and Technology, Wuhan University, Wuhan 430072, China

^b Department of Physics and Astronomy, and Wright Center for Photovoltaics Innovation and Commercialization, The University of Toledo, Toledo, OH 43606, USA

^c School of Electronic Science and Engineering, Southeast University, Nanjing 210096, China

ARTICLE INFO

Keywords:

Perovskite solar cells
Compositional and morphological engineering
Hysteresis
Flexible solar cells

ABSTRACT

We report on compositional and morphological engineering of mixed methylammonium (MA) and formamidinium (FA) lead triiodide ($\text{MA}_{1-x}\text{FA}_x\text{PbI}_3$) perovskite absorber layers to produce highly efficient planar and flexible perovskite solar cells (PVSCs) with reduced hysteresis. Incorporation of FA into the MAPbI_3 extends the absorption edge of the perovskite to longer wavelengths, leading to enhanced photocurrent of the resultant PVSCs. Moreover, adding a small amount of lead thiocyanate ($\text{Pb}(\text{SCN})_2$) additive into mixed perovskite precursor solutions significantly enlarges the grain size and prolongs the carrier lifetime, leading to improved device performance. With optimal compositional and morphological engineering, the average power conversion efficiency (PCE) improves from $15.74 \pm 0.74\%$ for pure MAPbI_3 PVSCs to $19.40 \pm 0.32\%$ for $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ PVSCs with 3% $\text{Pb}(\text{SCN})_2$ additive, exhibiting a high reproducibility and small hysteretic behavior. The best PVSC achieves a PCE of 20.10 (19.85)% measured under reverse (forward) voltage scan. Furthermore, the compositional and morphological engineering allowed the fabrication of efficient flexible PVSCs on indium-doped SnO_2 (ITO)/polyethylene terephthalate (PET) substrates, with the best PCE of 17.96 (16.10)% with a V_{OC} of 1.076 (1.020) V, a J_{SC} of 22.23 (22.23) mA/cm^2 and a FF of 75.10 (71.02)% when measured under reverse (forward) voltage scan. Our approach provides an effective pathway to fabricate highly efficient and reproducible planar PVSCs.

1. Introduction

Organic-inorganic metal halide perovskite solar cells (PVSCs) have gained tremendous attention in the past few years, [1–14] and the record power conversion efficiency (PCE) has recently exceeded 22%. [5–7] The bandgap (E_g) of the most commonly studied organic-inorganic perovskite absorber material, methylammonium lead triiodide (MAPbI_3) with $E_g \sim 1.60$ eV, is slightly higher than the optimal E_g range (1.2–1.45 eV) for single junction solar cells per the Shockley - Queisser limit [15,16]. A relatively smaller E_g is desirable for boosting the performance by increasing the portion of photons absorbed, thereby increasing the photogenerated current. Formamidinium lead triiodide (FAPbI_3) perovskite exhibits an E_g of ~ 1.45 eV, [17–20] which is much more suitable for single junction solar cells. However, FAPbI_3 perovskite films often contain a small portion of the undesirable yellow

phase (δ -phase) [7,21,22], which adversely affects the cell performance. Combining different cations has been reported recently to reduce the E_g of perovskite absorbers [4,5,20,21,23–25]. Among these methods, mixing MA and FA cations has shown promising results [5,8,26], but it is challenging to obtain good device reproducibility due to the facile formation of the undesirable δ -phase of FAPbI_3 . [7,21] Mixing MAPbI_3 with FAPbI_3 precursors has been proven to be an effective way to fabricate mixed FA and MA planar PVSCs [26]. However, a significant drawback of this approach is that the resultant ($\text{MA}_{1-x}\text{FA}_x\text{PbI}_3$) perovskite thin films contain rather small grains, hindering the use of relatively thick perovskite absorbers which is needed for maximum light absorption. On the other hand, the high annealing temperature of 150 °C for pure FAPbI_3 hinders its application on flexible substrates, while mixing MA and FA will lower the formation temperature to 100 °C which is more compatible for flexible

* Corresponding authors.

E-mail addresses: dewei_zhao@hotmail.com (D. Zhao), xzzhao@whu.edu.cn (X. Zhao), Yanfa.Yan@utoledo.edu (Y. Yan).

<http://dx.doi.org/10.1016/j.nanoen.2017.03.048>

Received 4 January 2017; Received in revised form 22 March 2017; Accepted 26 March 2017

Available online 28 March 2017

2211–2855/ © 2017 Published by Elsevier Ltd.

PVSCs.

High-quality perovskite film is a requisite for fabricating high-efficiency PVSCs. [27–29] PVSCs with small grains and low crystallinity typically exhibit poor fill factors (FF's), because absorber layers exhibit high density of grain boundaries, which could cause charge recombination [30,31]. Therefore, it is highly desirable to reduce the density of grain boundaries [26,31,32] and/or passivate the grain boundaries with other materials, such as (6,6)-phenyl-C₆₁-butyric acid methyl ester (PCBM) and PbI₂ [33–35]. Various works have been reported to improve the performance of PVSCs, including perovskite film quality improvement and interfacial engineering of the devices. [36–43] Recently, lead thiocyanate (Pb(SCN)₂) additives have been employed in MAPbI₃ and FA_{1-x}Cs_xPbI₃ perovskite films to enlarge the grains and increase the perovskite crystallinity [24,44]. It was found that Pb(SCN)₂ additives can significantly increase the grain size of perovskite films from about 200 nm to up to several micrometers, and the grain boundaries are passivated with excessive PbI₂ [24,44].

Here, we demonstrate that Pb(SCN)₂ additives can also significantly enlarge the grain size of mixed MA and FA lead triiodide (MA_{1-x}FA_xPbI₃) perovskite thin films and lead to highly efficient planar PVSCs. Mixtures of MA and FA cations extend the absorption edge to longer wavelengths as compared to pure MAPbI₃, leading to enhanced short-circuit current densities (J_{SC} 's). Adding a small amount of Pb(SCN)₂ additive into perovskite precursor remarkably increases the grain size of perovskite thin film and thus prolongs the carrier lifetime, which is responsible for the significantly enhanced device performance. With optimal compositional and morphological engineering, the averaged PCE of our PVSCs increases from 15.74 ± 0.74% for pure MAPbI₃ PVSCs (0 wt% Pb(SCN)₂) cells to 19.40 ± 0.32% for MA_{0.7}FA_{0.3}PbI₃ PVSCs with 3 wt% Pb(SCN)₂ additive. The best performing PVSC shows a PCE of 20.10 (19.85) % with an open-circuit voltage (V_{OC}) of 1.121 (1.119) V, a J_{SC} of 22.85 (22.86) mA/cm² and a FF of 78.46 (77.60) % when measured under reverse (forward) voltage scan, respectively. Furthermore, the compositional and morphological engineering allowed the fabrication of efficient flexible PVSCs on indium-doped SnO₂ (ITO)/polyethylene terephthalate (PET) substrates, with the best PCE of 17.96 (16.10)% with a V_{OC} of 1.076 (1.020) V, a J_{SC} of 22.23 (22.23) mA/cm² and a FF of 75.10 (71.02)% when measured under reverse (forward) voltage scan. Our results demonstrate that Pb(SCN)₂ additives can be used to effectively increase the crystal quality of the mixed cation perovskite absorbers and improve the photovoltaic performance of planar PVSCs with the regular cell configuration.

2. Experimental section

2.1. Preparation of ESLs

FTO glass (Pilkington, NSG TEC-15) was cleaned with detergent, facilitated by ultrasonication in water, acetone and iso-propanol baths successively before deposition of SnO₂ ESLs. The SnO₂ layer was deposited on the FTO substrates as reported previously.[45] Plasma-enhanced atomic layer deposition (PEALD) process was done with an equipment of Ensure Scientific Group AutoALD-PE V2.0 equipped with a plasma generator. Tetrakis(dimethylamino)-tin(IV) (99%, TDMA-Sn, Strem Chemicals Inc.) was used as the Sn precursor. Oxygen and Argon are used as oxidizer and carrier gases, respectively. The temperature of the reaction is fixed at 100 °C during the deposition process.

2.2. Perovskite precursor preparation

Lead iodide (PbI₂, Alfa Aesar, 99.9985%), methylammonium iodide (MAI, Dyesol), formamidinium iodide (FAI, Dyesol), lead thiocyanate (Pb(SCN)₂, Sigma-Aldrich, 99.5%), dimethyl sulfoxide (DMSO, Sigma-Aldrich, 99.8%) and N,N-dimethylformamide (DMF, Sigma-Aldrich, 99.8%) are purchased and used without further purification. The

perovskite precursor was prepared using a Lewis acid-base adduct approach with the mixture of MAI, FAI, PbI₂, and DMSO in DMF, where the molar ratio of DMSO and PbI₂ is 1:1 [46,47]. A 45% by weight precursor solution of MAPbI₃ was prepared with PbI₂, MAI and DMSO (molar ratio=1:1:1) in DMF. The solution was stirred for 12 h on a 60 °C hot plate before deposition. The FAPbI₃ precursor solution was made in the same process. The MA_{1-x}FA_xPbI₃ precursor was prepared by mixing two different solutions together with different volume ratio, where the ratio of MAI and FAI is changed from 1:0 to 0:1 (x=0, 0.2, 0.3, 0.4, 0.6, 0.8, 1.0). The resulting mixed precursor solution was stirred for one hour on a 60 °C hotplate and then purified using a 0.45 μm filter before spin-coating.

The precursors with Pb(SCN)₂ additives are prepared with fixed amounts of Pb(SCN)₂ with respect to the weight of PbI₂, where the content of Pb(SCN)₂ ranging from 0% to 5%.

2.3. Fabrication of PVSCs

C₆₀-SAM was deposited on PEALD SnO₂ as previously reported [45]. The perovskite precursor solution was spin-coated on the ESL first at 500 rpm for 3 s, and then at 4000 rpm for 60 s using a fast deposition-crystallization technique with diethyl ether as the anti-solvent agent. After spin coating, the perovskite film was annealed at 65 °C for 2 min and then 100 °C for 5 min. All of these processes were carried out in a N₂ filled glove box.

2,2',7,7'-tetrakis(N,N'-di-*p*-methoxyphenylamine)-9,9'-spirobi-fluorene (Spiro-OMeTAD) was used as the HSL and deposited on the perovskite film at 2000 rpm for 60 s. The Spiro-OMeTAD was co-doped using Co-TFSI and Li-TFSI. The Spiro-OMeTAD solution was prepared by dissolving 72.3 mg Spiro-OMeTAD (Shenzhen Feiming Science and Technology Co., Ltd., 99.0%) in 1 mL chlorobenzene (CB) with 28 μL 4-*tert*-butylpyridine (TBP) (Sigma-Aldrich, 96%), 18 μL Li-bis-(trifluoromethanesulfonyl) imide (Li-TFSI) (Sigma-Aldrich, 99.95%) (520 mg/mL in acetonitrile) and 18 μL Co(II)-TFSI salt (FK102, Dyesol) (300 mg/mL in acetonitrile). A layer of 80 nm gold (Au) was then deposited on the top of Spiro-OMeTAD using thermal evaporation. The working area of the devices was 0.08 cm² as defined by a shadow mask during the Au evaporation.

2.4. Characterizations

J-V curves were measured using a Keithley2400 sourcemeter under standard AM 1.5 G illumination using solar simulator (PV Measurements Inc.) with an output intensity of 100 mW/cm². For light intensity dependence test, the light intensity was later adjusted between 1 and 100 mW/cm² using neutral density filters. EQE measurement was carried out with an EQE system (PV Measurements Inc.) using 100 Hz chopped monochromatic light ranging from 300 nm to 900 nm under otherwise near-dark test conditions. The top-view and cross section structures of perovskite films and PVSCs were characterized with a field emission SEM instrument (Hitachi S-4800). Crystallinity and the crystal structure of the perovskite layer were analyzed with an Ultima III X-ray Diffractometer using a Ni-filtered Cu K_α x-ray source (Rigaku Corp.). Absorbance spectra were obtained with a UV-vis spectrophotometer (PerkinElmer Lambda 1050). Sheet resistance was measured using four-point probe method resistivity test system (PRO4-440N, Lucas labs). For steady-state photoluminescence (PL) and time resolved photoluminescence (TRPL) measurements, perovskite films were fabricated by spin-coating perovskite precursors on glass substrates followed by coating another encapsulating layer of polymethylmethacrylate (PMMA) after thermal annealing. PL measurements were performed in ambient air at room temperature. Samples were illuminated through the film side. A 532 nm continuous-wave laser (beam diameter ≈90 μm) at 40 mW/cm² was used as an excitation wavelength for steady-state PL measurement. PL signal was detected via Symphony-II CCD (from Horiba)

detector after a 300 g mm^{-1} grating monochromator (Integration time = 0.5 s). For TRPL measurements, samples were excited by a 532 nm pulsed laser (pulse width = 5 ps, beam diameter $\approx 150 \mu\text{m}$) at 10^{10} photons pulse $^{-1} \text{ cm}^{-2}$. TRPL measurements were performed with time correlated single photon counting (TCSPC) module (Becker & Hickel Simple Tau SPCM 130-E/M module) and the radiative recombination events were detected (Integration time = 600 s) via hybrid APD/PMT module (R10467U-50). PL decays of perovskite films with $x=0.3, 0.4, 0.6$ and 0.8 are single exponential while bi-exponential PL decays are observed in case of $x=0, 0.2$ and 1.0 . In case of bi-exponential PL decay, the photoluminescence intensity contribution of each component is proportional to the product of amplitude (A_i) and lifetime (τ_i). Therefore, the intensity average lifetime of bi-exponential PL decay is calculated as

$$\text{Mean lifetime}(\tau) = \frac{A_1\tau_1^2 + A_2\tau_2^2}{A_1\tau_1 + A_2\tau_2}$$

3. Results and discussion

The $\text{MA}_{1-x}\text{FA}_x\text{PbI}_3$ perovskite precursors were prepared by adding a FAPbI_3 precursor solution into a MAPbI_3 precursor with different molar ratios. The composition of a perovskite film is defined by the stoichiometry of mixed precursors.[5] The anneal temperatures of the $\text{MA}_{1-x}\text{FA}_x\text{PbI}_3$ perovskite films are 100°C for $x < 0.4$, 130°C for $x=0.6$ and 0.8 , 150°C for pure FAPbI_3 . A slightly higher annealing temperature is used for the film with a higher FA content, because FAPbI_3 has a yellow-to-black (δ to α) phase transition when heated to temperatures of around 150°C [17,18,48]. The steady-state photoluminescence (PL) spectra of the perovskite film with different compositions are shown in Fig. 1(a). As the FA content increases, the emission peak shows a red shift from 775 to 814 nm, which confirms the alloying of FAPbI_3 and MAPbI_3 . The E_g calculated from the PL emission peak position decreases linearly with the concentration increase of FA in MAPbI_3 (Fig. 1(c)), consistent with previous work reported in literature.[23] In addition, as shown in Fig. S1, the absorption onset of perovskite films shifts from 775 nm to 814 nm as observed from UV–vis spectroscopic measurements, in good agreement with the E_g trend obtained from the PL results. It is worth noting that the mixed MA and FA perovskite films show higher absorption compared to pure MAPbI_3 , which is beneficial for the enhancement of photogenerated currents.

In order to evaluate the effect of MA and FA alloy on the charge carrier lifetime, time resolved photoluminescence (TRPL) measurements were conducted to extract the photogenerated carrier lifetimes. Fig. 1(b) shows the normalized PL decays of perovskite films with various FA contents. PL decays of perovskite films with $x=0.3, 0.4, 0.6$ and 0.8 are single exponential indicating the absence of surface or interface recombination while bi-exponential PL decays are observed in case of $x=0, 0.2$ and 1.0 . The mean lifetime first increases and reaches a maximum value of 708 ns at $x=0.4$, suggesting a prolonged carrier lifetime with the incorporation of FA. However, further increase in the FA content 60% ($x > 0.6$) leads to the decrease in carrier lifetime, indicating the increased charge recombination, possibly due to the decreased film crystallinity (see discussions below). The TRPL results are summarized in Table S1.

X-ray diffraction (XRD) was carried out to evaluate the crystallinity of the mixed MA and FA perovskite films, as shown in Fig. 1(d). The (110) peak position shifts to lower angles with higher FA content, shown in Fig. 1(e), corresponding to the increased lattice constants after FA incorporation. This confirms the successful incorporation of FA cation into the perovskite crystal lattice, since the FA cation has a relatively larger ionic radius ($1.9\text{--}2.2 \text{ \AA}$) than MA (1.8 \AA).[5] The full width half maximum (FWHM) of the (110) diffraction peak increases monotonically with increasing FA content. Typically, decreased grain size and/or increased density of defects (lower crystallinity) are

responsible for the increase of FWHM. Because the average grain size increase as the FA content increases, the increased FWHM indicates decreased film crystallinity, which is likely attributable to the fact that the pure α -phase of FAPbI_3 is thermally unstable and it can easily revert to the non-perovskite δ -phase at ambient temperature, thus disrupting the single-phase crystallinity [5,26]. Therefore, pure FAPbI_3 perovskite is not ideal for fabricating highly efficient PVSCs due to the lower crystallinity despite having a more suitable E_g value. It is known that alloying FAPbI_3 with relatively stable perovskite materials, such as MAPbBr_3 [5,8,49] and CsPbI_3 [8,50] can stabilize the desirable α -phase of FAPbI_3 , which may reinforce the observation that the films with increasing MA content ($x > 0.2$) appear to have improved crystallinity than the more FA-rich films. Although the $\text{MA}_{1-x}\text{FA}_x\text{PbI}_3$ perovskite films exhibit improved light absorption compared to MAPbI_3 perovskite films, their grain sizes are smaller, as shown in Fig. 1(g), (h) and Fig. S2. The averaged grain size of the perovskite films with FA content less than 0.8 appears to be similar, around 100–200 nm. This is possibly due to the formation of Lewis acid–base adduct of $\text{FAI-PbI}_2 \cdot \text{DMSO}$, which is relatively weaker than $\text{MAI-PbI}_2 \cdot \text{DMSO}$ [46,47]. Therefore, the grain size is relatively small in the mixed MA and FA perovskite films. Small grains would lead to thin films with high density of grain boundaries, which could exhibit high charge recombination [31]. For pure FAPbI_3 , although the grain size is large, the film is very rough, likely due to the higher nucleus formation energy of FAPbI_3 , leading to grains with large size but lower crystallinity [26]. Therefore, it is highly desirable to increase the grain size and alleviate the charge recombination in the $\text{MA}_{1-x}\text{FA}_x\text{PbI}_3$ perovskite films in order to further improve the PCEs of PVSCs.

We fabricated PVSCs using $\text{MA}_{1-x}\text{FA}_x\text{PbI}_3$ absorbers with a planar regular device structure of fluorine doped SnO_2 (FTO)/ $\text{SnO}_2/\text{C}_{60}$ -SAM/ $\text{MA}_{1-x}\text{FA}_x\text{PbI}_3$ perovskite/spiroOMeTAD/Au[51–53], where C_{60} -SAM is C_{60} -self-assembled-monolayer and Spiro-OMeTAD is 2,2',7,7'-tetrakis(N,N'-di-*p*-methoxyphenylamine)-9,9'-spirobifluorene. The photovoltaic performances of the $\text{MA}_{1-x}\text{FA}_x\text{PbI}_3$ PVSCs are summarized in Fig. 2 and Table S2. The V_{OC} of the PVSCs decreases, but the J_{SC} of the PVSCs increases as the FA content increases due to the decreased E_g , except for the case for $x=1$. From the photographs of these perovskite films (Fig. S3), it is seen that the pure FAPbI_3 film is obstructed from the backside (glass side), and therefore the illuminated light is blocked, resulting in the reduction in the light harvesting and the low photocurrent. The FF slightly increases when x increase from 0 to 0.4, likely due to the prolonged carrier lifetime. However, when x is larger than 0.6, the FF deteriorates to below 70%. This may be due to the poor crystallinity and short carrier lifetimes of these perovskite films. As a result, the optimal device performance is obtained at $x=0.3$ ($\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$) due to both increased absorption and prolonged charge carrier lifetime, generating an average PCE of $17.61 \pm 0.32\%$. However, when the FA content is above 60% ($x > 0.6$), the device performance decreases dramatically due to the significant drop in V_{OC} and FF, likely due to the poor film morphology and low crystallinity as discussed above. The representative J - V curves of the best-performing device for composition and the corresponding external quantum efficiencies (EQEs) are shown in Fig. 2(e) and (f).

Various approaches have been proposed to increase the grain size of perovskite thin films in order to improve the device performance [30,31]. Previously, we have used $\text{Pb}(\text{SCN})_2$ additives to enlarge the grain sizes of MAPbI_3 and $\text{FA}_{1-x}\text{Cs}_x\text{PbI}_3$ perovskites films to obtain improved device performances [24,44]. Herein, we find that the $\text{Pb}(\text{SCN})_2$ additives in combined MAPbI_3 and FAPbI_3 precursor solutions can also effectively enlarge the grain size of $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ perovskite thin films. We incorporate various amounts of $\text{Pb}(\text{SCN})_2$ additives ranging from 0 to 5 wt% with respect to PbI_2 into the mixed precursors to investigate the effect of $\text{Pb}(\text{SCN})_2$ additives on the film morphology, crystalline, and device performance. We focused on the composition of $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ based on the above results.

Fig. 3(a) shows the absorbance spectra of $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ perovs-

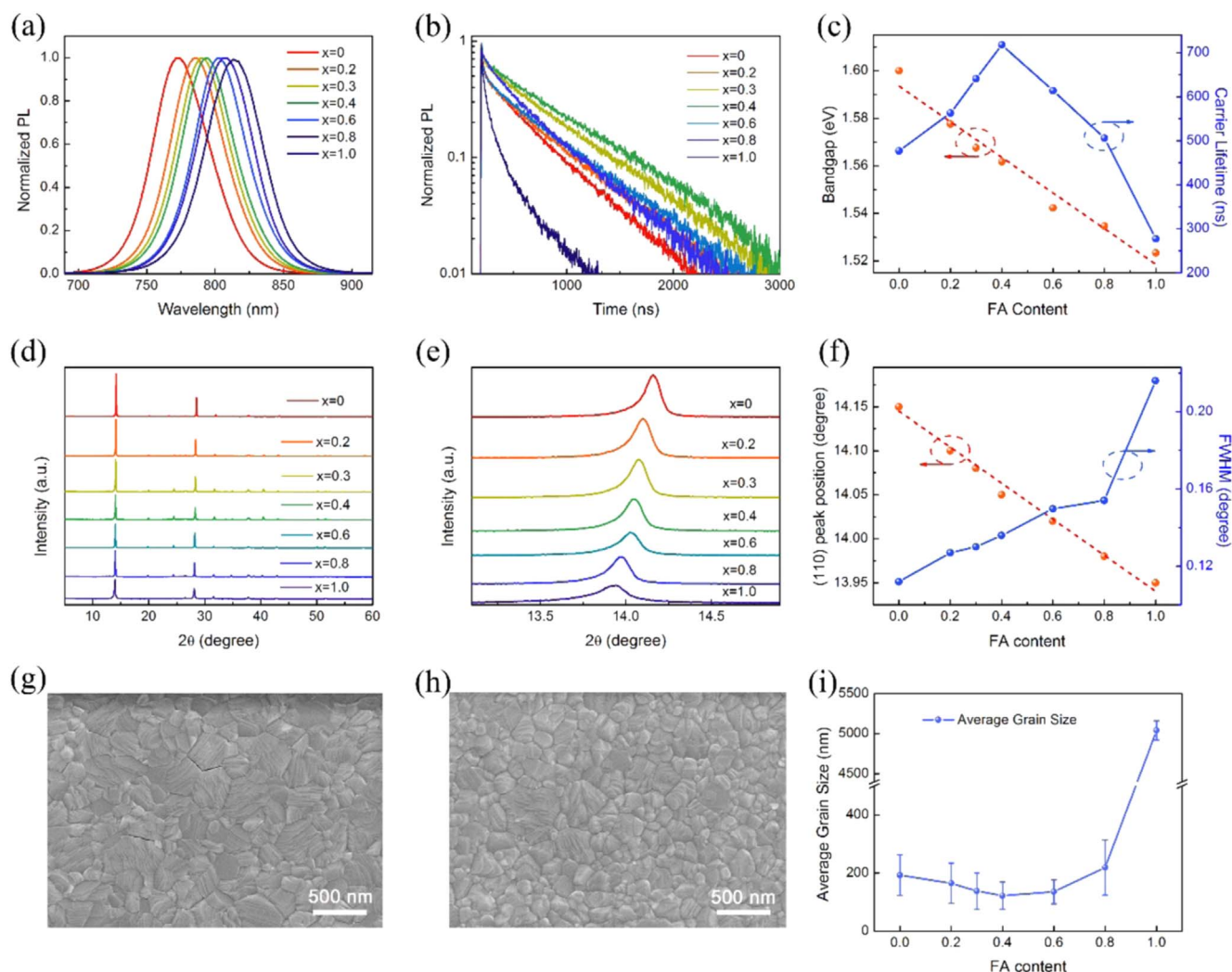


Fig. 1. Characterizations of $\text{MA}_{1-x}\text{FA}_x\text{PbI}_3$ films with x varying from 0 to 1. (a) Steady-state PL spectra and (b) PL decays of $\text{MA}_{1-x}\text{FA}_x\text{PbI}_3$ films. (c) Bandgap calculated from the PL emission peaks and average photogenerated carrier lifetimes. (d) XRD patterns and (e) magnified XRD patterns of the (110) diffraction peak. (f) (110) diffraction peak position from XRD and FWHM of (110) peak of $\text{MA}_{1-x}\text{FA}_x\text{PbI}_3$ films. Top-view SEM images of (g) MAPbI_3 and (h) $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ films. (i) Average grain size obtained from SEM images with different FA content.

kite films with various $\text{Pb}(\text{SCN})_2$ contents in the precursors, exhibiting identical absorption edges at around 798 nm. This indicates that the incorporation of $\text{Pb}(\text{SCN})_2$ additive in the precursors does not change the E_g 's of the perovskite absorbers, similar to our previous studies [24,44]. $\text{Pb}(\text{SCN})_2$ additive in the precursors also leads to $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ perovskite films with excessive PbI_2 locating at the grain boundaries, as verified by XRD and SEM. As shown in Fig. 3(b), the intensity of (110) diffraction peak at 14.2° increased dramatically compared to the $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ perovskite film without $\text{Pb}(\text{SCN})_2$, indicating higher crystallinity of the resulted films. Stronger PbI_2 peaks at 12.8° are observed for $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ perovskite films with higher $\text{Pb}(\text{SCN})_2$ concentrations, indicating increased PbI_2 in these perovskite films. The FWHM values of the (110) diffraction peak of perovskite films with various $\text{Pb}(\text{SCN})_2$ concentration are plotted in Fig. 3(c). It is seen that the FWHM becomes smaller first and reaches a minimum value at 3 wt%, and then increases. It can be inferred that the addition of $\text{Pb}(\text{SCN})_2$ will increase the crystallinity, which is due to the formation of large grain size with the addition of $\text{Pb}(\text{SCN})_2$ as discussed in our previous works [24,44]. However, if the amount of $\text{Pb}(\text{SCN})_2$ is too high, a large amount of PbI_2 will be present in the resultant perovskite films, which reduces the film phase purity and is detrimental for the device efficiency. By comparing the SEM images of perovskite films without (0 wt%) and with 3 wt% $\text{Pb}(\text{SCN})_2$, it is seen that the

average grain size is increased to 1760 nm for 3 wt% $\text{Pb}(\text{SCN})_2$ perovskite from 142 nm for 0 wt% $\text{Pb}(\text{SCN})_2$ perovskite, implying an increase of grain size by a factor of 12. The grains are much larger than the film thickness (about 500 nm) of our typical perovskites layers as shown in the cross-sectional SEM image (Fig. 3(f)), leading to less horizontally oriented grain boundaries that may block charge transportation [31]. A thin film with larger grains is expected to exhibit a lower carrier recombination rate than a thin film with smaller grains, resulting in an improved photovoltaic performance. The top-view SEM images of perovskite films with other different concentrations of $\text{Pb}(\text{SCN})_2$ are shown in Fig. S4. The average grain sizes are shown in Fig. 3(c), with the grain sizes increasing as the $\text{Pb}(\text{SCN})_2$ concentration increases. The grain size exceeds 2 μm at 5 wt% $\text{Pb}(\text{SCN})_2$ additive. However, the performance of PVSCs with 5 wt% $\text{Pb}(\text{SCN})_2$ additive drops largely due to presence of a large amount of excessive PbI_2 . Images of $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ films with various concentration of $\text{Pb}(\text{SCN})_2$ additive are shown in Fig. S5.

Several batches of PVSCs with different concentrations of $\text{Pb}(\text{SCN})_2$ were fabricated. The statistical representations of the resulting photovoltaic parameters, calculated from 234 devices, are shown in Fig. 4(a)–(d) and are also summarized in Table S3. The average V_{OC} first increases as the $\text{Pb}(\text{SCN})_2$ content increases and reaches a maximum value of 1.108 ± 0.008 V at 2 wt% $\text{Pb}(\text{SCN})_2$. The V_{OC} starts

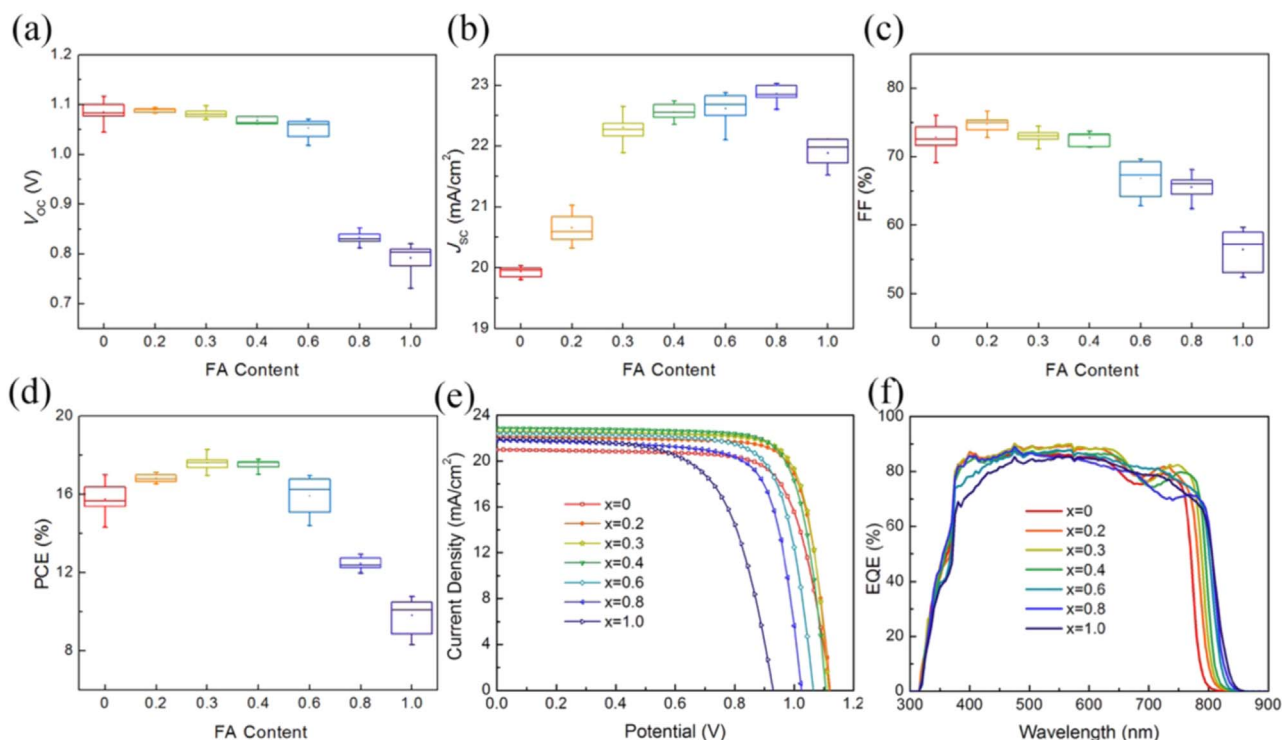


Fig. 2. Photovoltaic performance of PVSCs based on MA_{1-x}FA_xPbI₃ absorber with x varying from 0 to 1, showing the statistics of (a) V_{OC}, (b) J_{SC}, (c) FF and (d) PCE. (e) J-V curves and (f) EQE spectra of best-performing MA_{1-x}FA_xPbI₃ PVSCs.

to decrease when Pb(SCN)₂ content further increases, due to the larger amount of PbI₂ at the surface of the perovskite film, which will form a barrier between the perovskite and Spiro-OMeTAD layers. PbI₂ has a larger bandgap (~2.3 eV) [34] than perovskites, leading to relatively low conductivity and suppressing effective transfer of holes from the valence band of perovskite to the Spiro-OMeTAD. However, the J_{SC} does not change significantly and is maintained at around 21–23 mA/cm² regardless of the amount of Pb(SCN)₂. This is because the excess PbI₂ crystallites are mostly located at perovskite grain boundaries and

on film surfaces, rather than the light incident side, hence has no effect on light illumination. The FF increases from 76.25% to over 78% by increasing the Pb(SCN)₂ concentration, achieving a maximum value of 78.81 ± 0.91% at 3 wt%, and then decreases slightly when the Pb(SCN)₂ concentration further increases. As a result, the average PCE increases from 18.25 ± 0.50% for 0 wt% Pb(SCN)₂ cells to 19.40 ± 0.32% for the optimal 3 wt% devices. Higher concentrations of Pb(SCN)₂ (above 4 wt%) leads to reduced PCE of PVSCs, primarily due to the reduced V_{OC} although there is a slight decrease in the

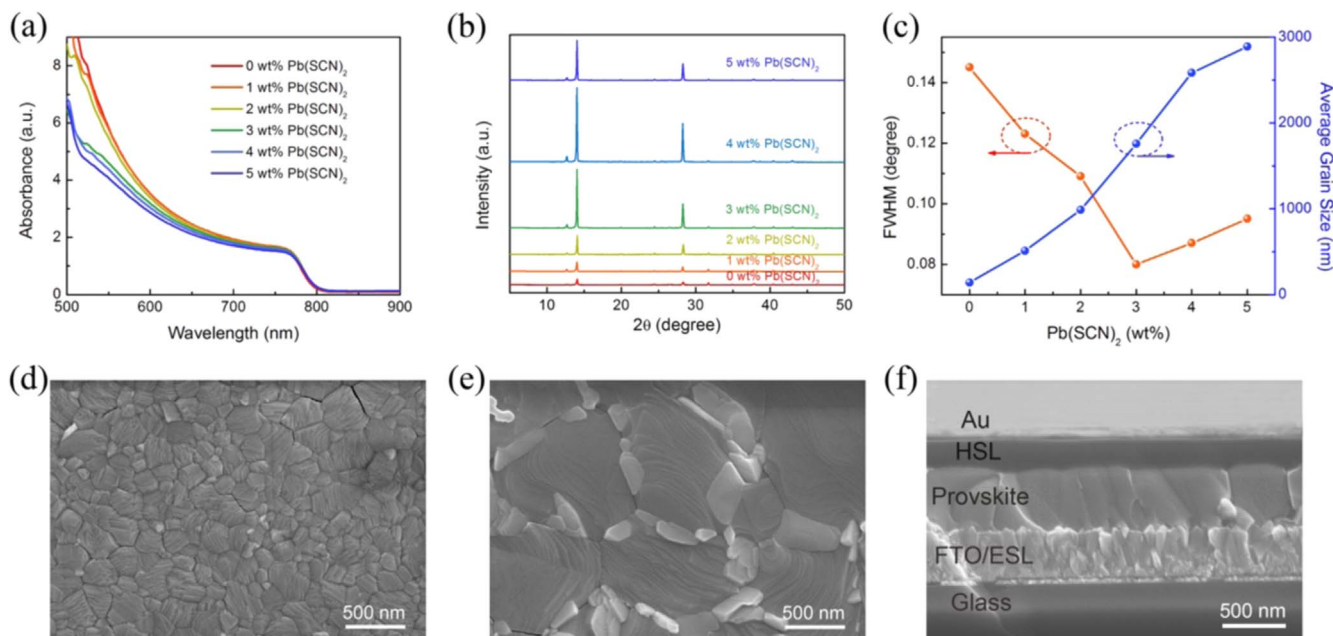


Fig. 3. Characterizations of MA_{0.7}FA_{0.3}PbI₃ films with various content of Pb(SCN)₂ additive. (a) UV-vis spectra, (b) XRD patterns, (c) FWHM of (110) diffraction peak and average grain size of MA_{0.7}FA_{0.3}PbI₃ films with various content of Pb(SCN)₂ additive. Plane-view SEM images of (d) MA_{0.7}FA_{0.3}PbI₃ and (e) MA_{0.7}FA_{0.3}PbI₃+3 wt% Pb(SCN)₂ films. (f) Cross-section SEM image of the planar structure PVSC.

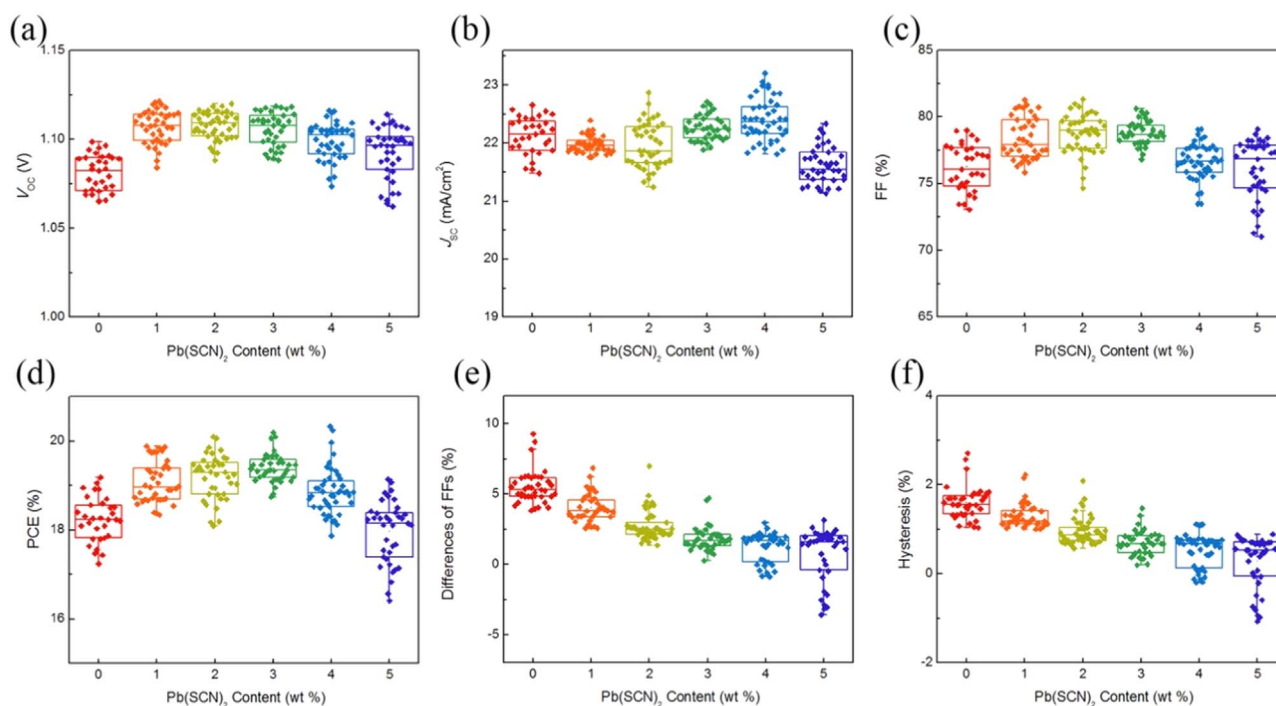


Fig. 4. Photovoltaic performance of $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ PVSCs with various concentrations of $\text{Pb}(\text{SCN})_2$ additives in the precursor. (a) V_{OC} , (b) J_{SC} , (c) FF, and (d) PCE measured under reverse voltage scans. The differences of (e) FFs and (f) PCEs (hysteresis) measured under reverse and forward scans.

highest $\text{Pb}(\text{SCN})_2$ concentration devices. On the other hand, it is worth mentioning that large perovskite grain sizes also reduce the observed J - V hysteresis [44]. The FF and PCE differences measured under reverse and forward voltage scans [54], are shown in Fig. 4(e) and (f), respectively. The FF and PCE differences (values measured under reverse voltage scan – values measured under forward voltage scan) decrease as the $\text{Pb}(\text{SCN})_2$ concentration increases, which can be ascribed to the increased perovskite grain size that leads to reduced density of grain boundaries and PbI_2 passivation of grain boundaries. Representative J - V curves of $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ PVSCs with various

amounts of $\text{Pb}(\text{SCN})_2$ additive are shown in Fig. S6.

Fig. 5 shows the comparison of the photovoltaic performance of PVSCs with absorbers of pure MAPbI_3 , $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ and $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ +3 wt% $\text{Pb}(\text{SCN})_2$ additive. Their representative J - V curves under reverse and forward bias scans with a scan rate of 1 V/s are shown in Fig. 5(a). The device with pure MAPbI_3 has a PCE of 16.49 (11.19) % with a V_{OC} of 1.102 (1.054) V, a J_{SC} of 19.99 (20.00) mA/cm^2 and a FF of 74.85 (53.07) % measured under reverse (forward) voltage scan. The PCE of the PVSC with $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ is increased to 18.87 (17.27) %, with a V_{OC} of 1.094 (1.075) V, a J_{SC} of

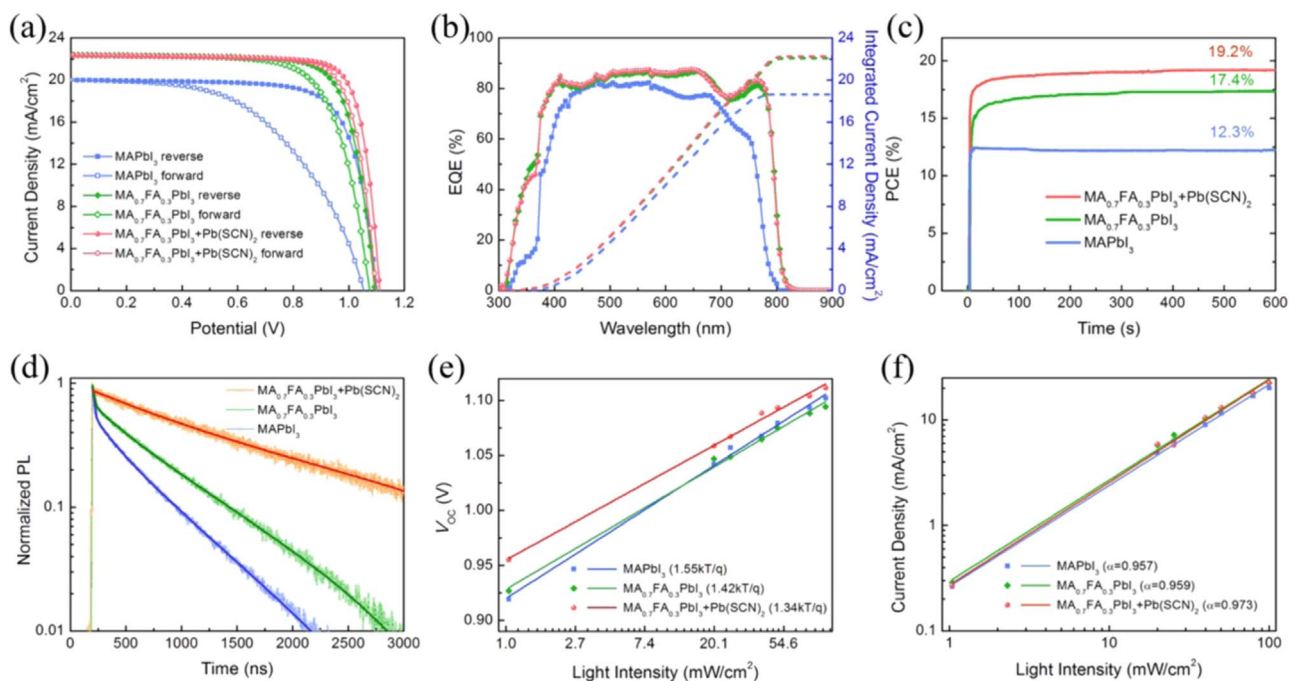


Fig. 5. (a) J - V curves, (b) EQE spectra and (c) stabilized PCEs of PVSCs based on MAPbI_3 , $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ and $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ +3 wt% $\text{Pb}(\text{SCN})_2$, respectively. (d) PL decays of three perovskite films. (e) V_{OC} and (f) J_{SC} versus light intensity of PVSCs with MAPbI_3 , $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ and $\text{MA}_{0.7}\text{FA}_{0.3}\text{PbI}_3$ +3 wt% $\text{Pb}(\text{SCN})_2$.

22.37 (22.37) mA/cm² and a FF of 77.11 (71.81) % measured under reverse (forward) scan. It is clear that the J_{SC} significantly increases due to the reduction in the Eg via FA substitution. With the Pb(SCN)₂ added into the mixed precursors, the MA_{0.7}FA_{0.3}PbI₃ PVSCs shows a V_{OC} of 1.112 (1.098) V, a J_{SC} of 22.34 (22.34) mA/cm² and a FF of 80.71 (78.87) %, generating a PCE of 20.03 (19.35)% under reverse (forward) scan direction. The integrated J_{SC} 's from the EQE spectra are 18.64, 22.08 and 22.24 mA/cm² for PVSCs with pure MAPbI₃, MA_{0.7}FA_{0.3}PbI₃ and MA_{0.7}FA_{0.3}PbI₃+3 wt% Pb(SCN)₂, respectively, which are in good agreement with the J_{SC} 's obtained from their J - V curves shown in Fig. 5(b). The PVSCs show increased stabilized power output with these alterations in the perovskite layer (Fig. 5(c)). The stabilized PCEs are 12.3%, 17.4% and 19.2% for the PVSCs with pure MAPbI₃, MA_{0.7}FA_{0.3}PbI₃ and MA_{0.7}FA_{0.3}PbI₃+3 wt% Pb(SCN)₂, respectively. The enhancement in V_{OC} and FF can be ascribed to the enlarged grain size (Fig. S4) and prolonged carrier lifetime, as obtained from TRPL results in Fig. 5(d). The average carrier lifetimes obtained from the fit are 470 ns, 635 ns and 1071 ns for pure MAPbI₃, MA_{0.7}FA_{0.3}PbI₃, and MA_{0.7}FA_{0.3}PbI₃+3 wt% Pb(SCN)₂, respectively. The dependence of V_{OC} and J_{SC} on light intensity ranging from 1 to 100 mW/cm² is shown in Fig. 5(e) and (f). The V_{OC} has a linear relationship with natural logarithmic light intensity, giving the fitted slopes of 1.55, 1.42 and 1.34 kT/q for PVSCs of pure MAPbI₃, MA_{0.7}FA_{0.3}PbI₃ and MA_{0.7}FA_{0.3}PbI₃+3 wt% Pb(SCN)₂, respectively. The results reveal the presence of Shockley–Read–Hall recombination in these PVSCs. However, the PVSC with MA_{0.7}FA_{0.3}PbI₃+3 wt% Pb(SCN)₂ shows lower trap-assisted recombination (lower slope) than the PVSC with MA_{0.7}FA_{0.3}PbI₃, which is still lower than for the PVSC with pure MAPbI₃. The power law dependence of J_{SC} on the light intensity shows a linear relation in a double logarithmic scale. The fitted slopes (α) of these three PVSCs are 0.957, 0.959 and 0.973 for PVSCs of pure MAPbI₃, MA_{0.7}FA_{0.3}PbI₃ and MA_{0.7}FA_{0.3}PbI₃+3 wt% Pb(SCN)₂, respectively. It has been reported that a solar cell with no space charge limited photocurrents will give a slope of 1 [55]. The PVSC with MA_{0.7}FA_{0.3}PbI₃+3 wt% Pb(SCN)₂ has more balanced charge carrier transportation than the pure MAPbI₃ and MA_{0.7}FA_{0.3}PbI₃ PVSCs due to the improved perovskite film quality. Fig. S7 shows cross-sectional SEM images of devices with three different absorbers, MAPbI₃, MA_{0.7}FA_{0.3}PbI₃, and MA_{0.7}FA_{0.3}PbI₃+3 wt% Pb(SCN)₂. The absorber layers have a very similar thickness of about 500 nm. The MAPbI₃ and MA_{0.7}FA_{0.3}PbI₃ thin films have small grains with sizes smaller than the film thickness. Therefore, the absorbers contain horizontal grain boundaries, presenting a significant risk for recombination of photogenerated electrons and holes when they travel through the absorber layer. However, the MA_{0.7}FA_{0.3}PbI₃+3 wt% Pb(SCN)₂ film shows much larger grains with sizes larger than the film thickness. The absence of horizontal grain boundaries should result in a much lower rate of carrier recombination, leading to improved device performances.

Our best-performing PVSC with MA_{0.7}FA_{0.3}PbI₃+3 wt% Pb(SCN)₂ shows a PCE of 20.10 (19.85) % with a V_{OC} of 1.121 (1.119) V, a J_{SC} of 22.85 (22.86) mA/cm² and a FF of 78.46 (77.60) % measured under reverse (forward) voltage scan, showing a very small degree of J - V hysteresis (Fig. 6(a)). The corresponding EQE spectrum and integrated current density under a standard AM 1.5 G spectrum are shown in Fig. 6(b). The EQE-integrated J_{SC} is 22.60 mA/cm², consistent with the J_{SC} obtained from the J - V curve. Fig. 6(c) shows the maximum power output of ~19.8% measured using the maximum power point tracking method. The power output of this PVSC is very stable under one sun illumination for 600 s without any isolation from the ambient environment, during which time no significant degradation was observed. The current density (~21.7 mA/cm²) hardly changed while the voltage increased slightly in the first several seconds under illumination then achieved a stable value of 0.91 V. The stable power output is consistent with the value obtained from J - V curve.

The low-temperature fabrication process is particularly beneficial

for making flexible PVSCs. We have fabricated flexible PVSCs on indium-doped SnO₂ (ITO)/polyethylene terephthalate (PET) substrates. The compositional and morphological engineering conditions optimized for PVSCs fabricated on glass substrates are suitable for flexible PVSCs on PET substrates. The J - V curves under reverse and forward scans of the best-performing flexible PVSC are shown in Fig. 7(a), exhibiting a PCE of 17.96 (16.10)% with a V_{OC} of 1.076 (1.020) V, a J_{SC} of 22.23 (22.23) mA/cm² and a FF of 75.10 (71.02)% when measured under reverse (forward) voltage scan with a sweep rate of 1 V/s. The degrees of hysteresis of the flexible PVSCs are higher than the PVSCs fabricated on rigid FTO glass substrates. This is likely due to the relatively higher sheet resistance of the ITO/PET substrate (~45 Ω/□) compared to that of FTO glass substrate (~15 Ω/□). The integrated current density from the EQE spectrum of this flexible PVSC is 21.90 mA/cm² as shown in Fig. 7(b), in good agreement with the J_{SC} value obtained from J - V curve. The XRD pattern and SEM image of the MA_{0.7}FA_{0.3}PbI₃+3 wt% Pb(SCN)₂ perovskite film deposited on ITO/PET substrate are shown in Fig. 7(c), revealing the successful formation of high quality perovskite film. We have also measured the effect of bending fatigue cycles on device performance as shown in Fig. 7(d), (e) and (f). The J - V curve of flexible PVSC under reverse scan shows that FF decreases significantly with the increase number of bending, which can be ascribed to the dramatically increased sheet resistance of the flexible ITO/PET substrate as shown in Fig. 7(e), which leads to the increased series resistance of the PVSC. This is consistent with Fig. S8, showing no obvious change on the film morphology after 340 times bending. After 340 bending with a minimum radius of curvature of 5 mm, the PCE is maintained at 14.30%, approximately 80% of its initial PCE of 17.96%, indicating a good flexibility endurance of our flexible PVSCs. The image of our flexible PVSC is shown as the inset of Fig. 7(f). Therefore, higher performance flexible PVSCs are expected if flexible substrates with higher conductivity and better bending endurance become available [56].

4. Conclusion

We have employed compositional and morphological engineering of mixed MA_{1-x}FA_xPbI₃ perovskite absorber layers to improve the performance of planar PVSCs. The alloying of FA and MA extends the absorption edge of the perovskite absorbers to longer wavelengths, leading to enhanced J_{SC} . The introduction of a small amount of Pb(SCN)₂ into the perovskite precursors enlarges the grain size and prolongs the charge carrier lifetimes in the perovskite films, leading to significant improvement in V_{OC} 's and FFs in the resultant devices. With optimal compositional and morphological engineering, the average PCE improves from 15.74 ± 0.74% for pure MAPbI₃ PVSCs to 19.40 ± 0.32% for MA_{0.7}FA_{0.3}PbI₃ PVSCs with 3% Pb(SCN)₂ additive. Our best-performing planar MA_{0.7}FA_{0.3}PbI₃ PVSC (with 3% Pb(SCN)₂ additive) has achieved a PCE of 20.10 (19.85) % when measured under reverse (forward) voltage scan. The flexible PVSC shows a very high efficiency of 17.96% under reverse scan. Our results suggest that incorporating Pb(SCN)₂ additive in the precursors is an effective approach for fabricating highly efficient planar mixed cation PVSCs with reduced hysteresis and high reproducibility.

Acknowledgements

This work is financially supported by the U.S. Department of Energy (DOE) SunShot Initiative under the Next Generation Photovoltaics 3 program (DE-FOA-0000990), National Science Foundation under contract no. CHE-1230246 and DMR-1534686, and the Ohio Research Scholar Program. N. Shrestha thanks the U.S. Air Force Research Laboratory for support under contract # FA9453-11-C-0253. This work also received financial support from National Basic Research Program of China (2011CB933300), National Science Fund for Distinguished Young Scholars (50125309), National Natural

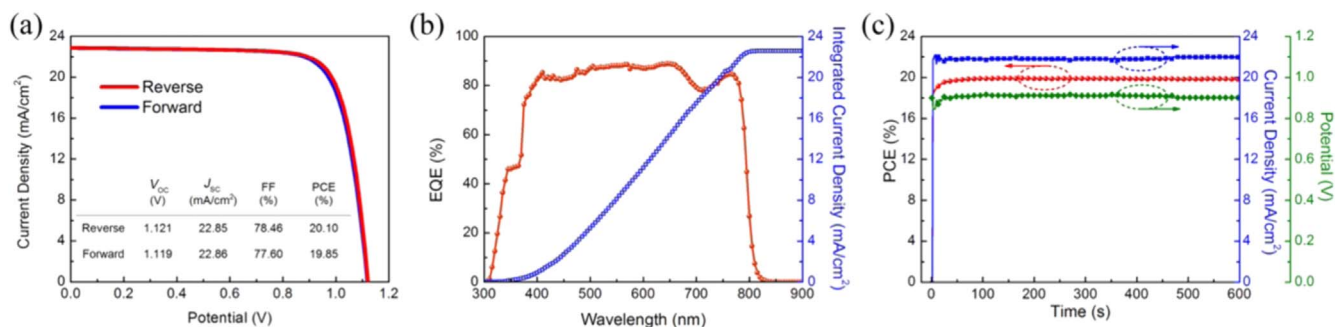


Fig. 6. (a) The J - V curves of the best-performing PVSC measured under reverse and forward scans, (b) the corresponding EQE spectrum and integrated photocurrent density. (c) PCE, photocurrent density and voltage measured using maximum power point tracking method.

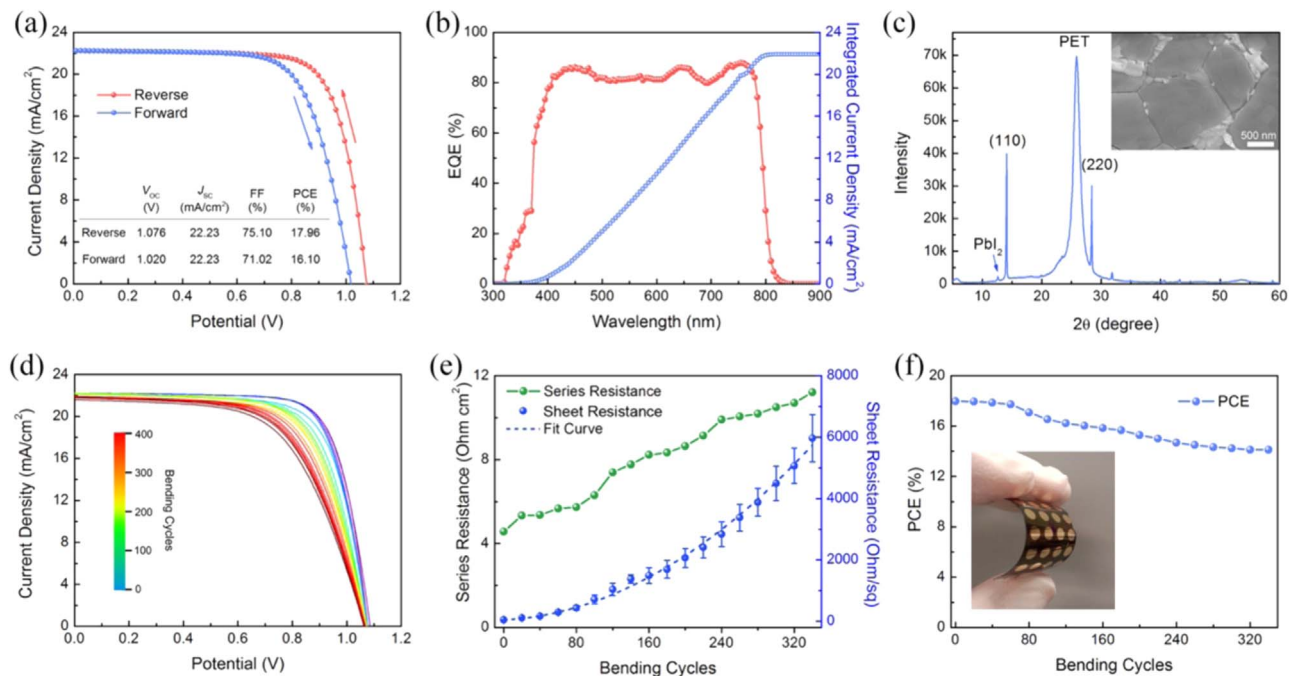


Fig. 7. (a) The J - V curves of the best-performing flexible PVSC measured under reverse and forward scans with a scan rate of 1 V/s, (b) the corresponding EQE spectrum and integrated photocurrent density of 21.90 mA/cm². (c) XRD pattern and top-view SEM image (inset) of perovskite films on ITO/PET substrate. (d) The J - V curves of the flexible PVSC with various bending cycles under reverse voltage scan. (e) The series resistance of the flexible PVSC calculated from the J - V curves and the sheet resistance of the ITO/PET substrate with various bending cycles. (f) PCE of the flexible PVSC with various bending cycles.

Science Foundation of China (Grants 51272184 and 91433203). J. Chen thanks the partially financial support from National Key R&D Program of China 2016YFB0401600, National Natural Science Foundation Project (61674029), and Natural Science Foundation Project of Jiangsu Province (BK20151417).

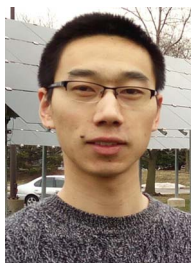
Appendix A. Supporting information

Supplementary data associated with this article can be found in the online version at doi:10.1016/j.nanoen.2017.03.048.

References

- [1] A. Kojima, K. Teshima, Y. Shirai, T. Miyasaka, *J. Am. Chem. Soc.* 131 (2009) 6050–6051.
- [2] M.M. Lee, J. Teuscher, T. Miyasaka, T.N. Murakami, H.J. Snaith, *Science* 338 (2012) 643–647.
- [3] M. Liu, M.B. Johnston, H.J. Snaith, *Nature* 501 (2013) 395–398.
- [4] D.P. McMeekin, G. Sadoughi, W. Rehman, G.E. Eperon, M. Saliba, M.T. Hörantner, A. Haghighirad, N. Sakai, L. Korte, B. Rech, M.B. Johnston, L.M. Herz, H.J. Snaith, *Science* 351 (2016) 151–155.
- [5] N.J. Jeon, J.H. Noh, W.S. Yang, Y.C. Kim, S. Ryu, J. Seo, S.I. Seok, *Nature* 517 (2015) 476–480.
- [6] W.S. Yang, J.H. Noh, N.J. Jeon, Y.C. Kim, S. Ryu, J. Seo, S.I. Seok, *Science* 348 (2015) 1234–1237.
- [7] M. Saliba, T. Matsui, K. Domanski, J.-Y. Seo, A. Ummadisingu, S.M. Zakeeruddin, J.-P. Correa-Baena, W.R. Tress, A. Abate, A. Hagfeldt, M. Grätzel, *Science* 354 (2016) 206–209.
- [8] D. Bi, W. Tress, M.I. Dar, P. Gao, J. Luo, C. Renevier, K. Schenk, A. Abate, F. Giordano, J.-P.C. Baena, *Sci. Adv.* 2 (2016) e1501170.
- [9] A. Mei, X. Li, L. Liu, Z. Ku, T. Liu, Y. Rong, M. Xu, M. Hu, J. Chen, Y. Yang, *Science* 345 (2014) 295–298.
- [10] W. Liao, D. Zhao, Y. Yu, C.R. Grice, C. Wang, A.J. Cimaroli, P. Schulz, W. Meng, K. Zhu, R.-G. Xiong, Y. Yan, *Adv. Mater.* 28 (2016) 9333–9340.
- [11] W. Liao, D. Zhao, Y. Yu, N. Shrestha, K. Ghimire, C.R. Grice, C. Wang, Y. Xiao, A.J. Cimaroli, R.J. Ellingson, N.J. Podraza, K. Zhu, R.-G. Xiong, Y. Yan, *J. Am. Chem. Soc.* 138 (2016) 12360–12363.
- [12] Z. Yu, B. Chen, P. Liu, C. Wang, C. Bu, N. Cheng, S. Bai, Y. Yan, X. Zhao, *Adv. Funct. Mater.* 26 (2016) 4866–4873.
- [13] Z. Song, A. Abate, S.C. Watthage, G.K. Liyanage, A.B. Phillips, U. Steiner, M. Graetzel, M.J. Heben, *Adv. Energy Mater.* 6 (2016) 1600846.
- [14] B. Chen, Y. Bai, Z. Yu, T. Li, X. Zheng, Q. Dong, L. Shen, M. Boccard, A. Gruverman, Z. Holman, J. Huang, *Adv. Energy Mater.* 6 (2016) 1601128.
- [15] W. Shockley, H.J. Queisser, *J. Appl. Phys.* 32 (1961) 510–519.
- [16] M.A. Green, *Prog. Photo.: Res. Appl.* 20 (2012) 472–476.
- [17] S. Pang, H. Hu, J. Zhang, S. Lv, Y. Yu, F. Wei, T. Qin, H. Xu, Z. Liu, G. Cui, *Chem. Mater.* 26 (2014) 1485–1491.
- [18] G.E. Eperon, S.D. Stranks, C. Menelaou, M.B. Johnston, L.M. Herz, H.J. Snaith, *Energy Environ. Sci.* 7 (2014) 982–988.
- [19] F. Wang, H. Yu, H. Xu, N. Zhao, *Adv. Funct. Mater.* 25 (2015) 1120–1126.
- [20] S. Aharon, A. Dymshits, A. Rotem, L. Etgar, *J. Mater. Chem. A* 3 (2015) 9171–9178.
- [21] M. Saliba, T. Matsui, J.-Y. Seo, K. Domanski, J.-P. Correa-Baena,

- M.K. Nazeeruddin, S.M. Zakeeruddin, W. Tress, A. Abate, A. Hagfeldt, M. Gratzel, *Energy Environ. Sci.* 9 (2016) 1989–1997.
- [22] Z. Song, S.C. Watthage, A.B. Phillips, B.L. Tompkins, R.J. Ellingson, M.J. Heben, *Chem. Mater.* 27 (2015) 4612–4619.
- [23] Z. Yang, C.-C. Chueh, P.-W. Liang, M. Crump, F. Lin, Z. Zhu, A.K.Y. Jen, *Nano Energy* 22 (2016) 328–337.
- [24] Y. Yu, C. Wang, C.R. Grice, N. Shrestha, J. Chen, D. Zhao, W. Liao, A.J. Cimaroli, P.J. Roland, R.J. Ellingson, Y. Yan, *ChemSusChem* 9 (2016) 3288–3297.
- [25] Z. Li, M. Yang, J.-S. Park, S.-H. Wei, J.J. Berry, K. Zhu, *Chem. Mat.* 28 (2016) 284–292.
- [26] Y. Deng, Q. Dong, C. Bi, Y. Yuan, J. Huang, *Adv. Energy Mater.* 6 (2016) 1600372.
- [27] D. Zhao, W. Ke, C.R. Grice, A.J. Cimaroli, X. Tan, M. Yang, R.W. Collins, H. Zhang, K. Zhu, Y. Yan, *Nano Energy* 19 (2016) 88–97.
- [28] D. Zhao, M. Sexton, H.Y. Park, G. Baure, J.C. Nino, F. So, *Adv. Energy Mater.* 5 (2015) 1401855.
- [29] W. Li, J. Fan, Y. Mai, L. Wang, *Adv. Energy Mater.* 7 (2016) 1601433.
- [30] W. Nie, H. Tsai, R. Asadpour, J.-C. Blancon, A.J. Neukirch, G. Gupta, J.J. Crochet, M. Chhowalla, S. Tretiak, M.A. Alam, H.-L. Wang, A.D. Mohite, *Science* 347 (2015) 522–525.
- [31] Z. Xiao, Q. Dong, C. Bi, Y. Shao, Y. Yuan, J. Huang, *Adv. Mater.* 26 (2014) 6503–6509.
- [32] X. Meng, Y. Bai, S. Xiao, T. Zhang, C. Hu, Y. Yang, X. Zheng, S. Yang, *Nano Energy* 30 (2016) 341–346.
- [33] Y. Shao, Z. Xiao, C. Bi, Y. Yuan, J. Huang, *Nat. Commun.* 5 (2014) 5784.
- [34] Q. Chen, H. Zhou, T.-B. Song, S. Luo, Z. Hong, H.-S. Duan, L. Dou, Y. Liu, Y. Yang, *Nano Lett.* 14 (2014) 4158–4163.
- [35] X. Liu, F. Lin, C.-C. Chueh, Q. Chen, T. Zhao, P.-W. Liang, Z. Zhu, Y. Sun, A.K.Y. Jen, *Nano Energy* 30 (2016) 417–425.
- [36] S. Song, B.J. Moon, M.T. Hörantner, J. Lim, G. Kang, M. Park, J.Y. Kim, H.J. Snaith, T. Park, *Nano Energy* 28 (2016) 269–276.
- [37] Z.-L. Tseng, C.-H. Chiang, S.-H. Chang, C.-G. Wu, *Nano Energy* 28 (2016) 311–318.
- [38] X. Sun, C. Zhang, J. Chang, H. Yang, H. Xi, G. Lu, D. Chen, Z. Lin, X. Lu, J. Zhang, Y. Hao, *Nano Energy* 28 (2016) 417–425.
- [39] R. Fan, Y. Huang, L. Wang, L. Li, G. Zheng, H. Zhou, *Adv. Energy Mater.* 6 (2016) 1600460.
- [40] M. Li, Z.-K. Wang, Y.-G. Yang, Y. Hu, S.-L. Feng, J.-M. Wang, X.-Y. Gao, L.-S. Liao, *Adv. Energy Mater.* 6 (2016) 1601156.
- [41] Z. Li, J. Tinkham, P. Schulz, M. Yang, D.H. Kim, J. Berry, A. Sellinger, K. Zhu, *Adv. Energy Mater.* 7 (2017) 1601451.
- [42] W. Ke, D. Zhao, C.R. Grice, A.J. Cimaroli, J. Ge, H. Tao, H. Lei, G. Fang, Y. Yan, *J. Mater. Chem. A* 3 (2015) 17971–17976.
- [43] D. Zhao, Y. Yu, C. Wang, W. Liao, N. Shrestha, C.R. Grice, A.J. Cimaroli, L. Guan, R.J. Ellingson, K. Zhu, X. Zhao, R.-G. Xiong, Y. Yan, *Nat. Energy* 2 (2017) 17018.
- [44] W. Ke, C. Xiao, C. Wang, B. Saparov, H.-S. Duan, D. Zhao, Z. Xiao, P. Schulz, S.P. Harvey, W. Liao, W. Meng, Y. Yu, A.J. Cimaroli, C.-S. Jiang, K. Zhu, M. Al-Jassim, G. Fang, D.B. Mitzi, Y. Yan, *Adv. Mater.* 28 (2016) 5214–5221.
- [45] C. Wang, D. Zhao, C.R. Grice, W. Liao, Y. Yu, A. Cimaroli, N. Shrestha, P.J. Roland, J. Chen, Z. Yu, P. Liu, N. Cheng, R.J. Ellingson, X. Zhao, Y. Yan, *J. Mater. Chem. A* 4 (2016) 12080–12087.
- [46] N. Ahn, D.-Y. Son, I.-H. Jang, S.M. Kang, M. Choi, N.-G. Park, *J. Am. Chem. Soc.* 137 (2015) 8696–8699.
- [47] J.-W. Lee, H.-S. Kim, N.-G. Park, *Acc. Chem. Res.* 49 (2016) 311–319.
- [48] J.-W. Lee, D.-J. Seol, A.-N. Cho, N.-G. Park, *Adv. Mater.* 26 (2014) 4991–4998.
- [49] J. Peng, T. Duong, X. Zhou, H. Shen, Y. Wu, H.K. Mulmudi, Y. Wan, D. Zhong, J. Li, T. Tsuzuki, K.J. Weber, K.R. Catchpole, T.P. White, *Adv. Energy Mater.* 7 (2017) 1601768.
- [50] J.-W. Lee, D.-H. Kim, H.-S. Kim, S.-W. Seo, S.M. Cho, N.-G. Park, *Adv. Energy Mater.* 5 (2015) 1501310.
- [51] K. Wojciechowski, S.D. Stranks, A. Abate, G. Sadoughi, A. Sadhanala, N. Kopidakis, G. Rumbles, C.-Z. Li, R.H. Friend, A.K.Y. Jen, H.J. Snaith, *ACS nano* 8 (2014) 12701–12709.
- [52] X. Liu, K.-W. Tsai, Z. Zhu, Y. Sun, C.-C. Chueh, A.K.Y. Jen, *Adv. Mater. Interfaces* 3 (2016) 1600122.
- [53] B. Roose, J.-P.C. Baena, K.C. Gödel, M. Graetzel, A. Hagfeldt, U. Steiner, A. Abate, *Nano Energy* 30 (2016) 517–522.
- [54] E.H. Anaraki, A. Kermanpur, L. Steier, K. Domanski, T. Matsui, W. Tress, M. Saliba, A. Abate, M. Gratzel, A. Hagfeldt, J.-P. Correa-Baena, *Energy Environ. Sci.* 9 (2016) 3128–3134.
- [55] V. Mihailetchi, J. Wildeman, P. Blom, *Phys. Rev. Lett.* 94 (2005) 126602.
- [56] J. Yoon, H. Sung, G. Lee, W. Cho, N. Ahn, H.S. Jung, M. Choi, *Energy Environ. Sci.* 10 (2016) 337–345.



Changlei Wang is a Ph.D. student in Wuhan University under the supervision of Prof. Xingzhong Zhao. Currently, he is working in Prof. Yanfa Yan's group as a visiting scholar at The University of Toledo. His research interest is perovskite solar cells and synthesis of nanomaterials.



Dewei Zhao is a Research Assistant Professor in Wright Center for Photovoltaics Innovation and Commercialization and Department of Physics and Astronomy at The University of Toledo. He received his Ph.D. degree in electrical engineering from Nanyang Technological University, Singapore in 2011. He joined Prof Yanfa Yan's group in 2014. His recent research is focused on vacuum-processed and solution-processed perovskite solar cells, including material synthesis, device fabrication/characterization, as well as organic optoelectronics such as solar cells and light-emitting diodes.



Yue Yu received her Bachelor of Science degree in Material Science and Engineering from the University of Science and Technology, Beijing in 2012. After graduating, she entered the University of Toledo for a Ph.D. degree in Physics with specialization in photovoltaics. She has broad research interests in thin film solar cells and photoelectrochemical water splitting. She is currently working on perovskite solar cell fabrications and characterizations.



Niraj Shrestha is a Ph.D. student at University of Toledo. He received his MS degree from University of Memphis, Memphis, TN in 2013. His current research interests include study of semiconductor thin films based on their static and dynamic optical properties including techniques of ultrafast pump-probe spectroscopy, steady state photoluminescence spectroscopy and time resolved photoluminescence spectroscopy.



Corey R. Grice earned Bachelor of Science in Engineering degrees in Chemical Engineering and Materials Science & Engineering from the University of Michigan in 2005. From 2005 until 2011 he worked in the private sector, primarily in fuel cell research and development. He entered graduate school at the University of Toledo in 2011 and completed a Professional Science Masters degree in Physics with specialization in Photovoltaics in 2013. He is currently pursuing a PhD in Physics under the guidance of Dr. Yanfa Yan, with research focusing on various novel and/or low cost thin film photovoltaic and photoelectrochemical materials and devices.



Weiqiang Liao received his Ph.D. degree in Materials Physics and Chemistry from Southeast University, China in 2017. He joined Prof Yanfa Yan's group in 2015 as a visiting scholar. His research interests include molecular ferroelectrics, organic-inorganic hybrid functional materials, and perovskite solar cells. He is currently working on ferroelectric photovoltaic solar cells.



Xing-Zhong Zhao is a Cheung Kong Professor of Physics at Wuhan University, China. He received his Ph.D. from University of Science and Technology Beijing in 1989. From 1990–1995, he was an Assistant and Associate Professor at Huazhong University of Science and Technology. He then worked as an Adjunct Associate Professor and post doctorate research fellow at Pennsylvania State University (1995–1998). He visited MIT as a senior visiting scientist in 2006 for six months. His research interests include the dye-sensitized and perovskite solar cells, microfluidic-based enriching and isolation of circulating tumor cells and fetal nucleated red blood cells for biomedical analysis and diagnosis and

related applications.



Alexander J Cimaroli received a B.S. in Applied Physics from Purdue University, West Lafayette, Indiana in 2010. He received his Ph.D. degree in the Physics department at the University of Toledo. His research focuses on thin film photovoltaic devices, device fabrication and characterization.



Yanfa Yan has been an Ohio Research Scholar Chair and Professor in the Department of Physics and Astronomy at The University of Toledo, since 2011. Previously, he was a Principal Scientist at the National Renewable Energy Laboratory. He earned his Ph. D. in Physics from Wuhan University in 1993. His expertise includes thin-film solar cell fabrication, defect physics of semiconductors, and nanoscale characterization of microstructures, interfaces, and defects in thin-film photovoltaic materials. He is a Fellow of the American Physical Society.



Jing Chen was born in China, in 1982. She received the B.S. degree in Chemistry from the Nanjing Normal University, China in 2001 and M.S. and the Ph.D. degrees in Physical Electronics from Southeast University, China in 2011. From 2011–2014, she was lecturer with Display and Research Center of Southeast University. Since 2014, she has been an Associated Professor with School of Electronic Science and Engineering, Southeast University. She is the author of more than 50 articles, and hosts more than 10 patents. Her research interests include quantum dot synthesis and application of opt-electric devices, such as quantum dot sensitized solar cell, quantum dot light emitting diodes, sensors.



Randy Ellingson, Professor of Physics and faculty member of the University of Toledo's Wright Center for Photovoltaics Innovation and Commercialization, studies photovoltaic materials, devices, and systems. Methods include ultrafast and steady-state optical spectroscopies, synthesis and solution-processing of nanocrystalline and quantum-confined semiconductors, device modeling, and PV system modeling and energy production forecasting. He received his B.A. in Physics from Carleton College and his Ph.D. degree in Applied Physics from Cornell University. Following 14 years at the National Renewable Energy Laboratory in Colorado, he joined the University of Toledo in 2008 to focus on the application of advanced

materials and characterization to commercially relevant solar energy technologies.